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<p>(21) International Application Number: PCT/US92/02645 (22) International Filing Date: 1 April 1992 (01.04.92) (30) Priority data: 679,843 3 April 1991 (03.04.91) US (71) Applicant: EASTMAN KODAK COMPANY [US/US]: 343 State Street, Rochester, NY 14650-2201 (US). (72) Inventors: FRANKE, Hans-George ; 683 Wilder Road, Hilton, NY 14468 (US). PRINCE, Eric, T. ; 24 Watchet Lane, Fairport, NY 14450 (US). (74) Agent: OWENS, Raymond, L. ; 343 State Street, Rochester, NY 14650-2201 (US).</p>		<p>(81) Designated States: AT (European patent), BE (European patent), CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, LU (European patent), MC (European patent), NL (European patent), SE (European patent). Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i></p>
<p>(54) Title: HIGH DURABILITY MASK FOR DRY ETCHING OF GaAs</p> <p>(57) Abstract</p> <p>A mask is described which enables the fabrication of features in GaAs such as waveguides, channels, facets, mesas, and mirrors by dry etch processing in chlorine containing ambients. The mask consists of an amorphous form of carbon which may contain incorporated hydrogen. The mask can be applied, patterned and removed through dry processing techniques.</p>		

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HIGH DURABILITY MASK FOR DRY ETCHING OF GaAs**Technical Field**

5 This invention relates to the protection of
desired regions of GaAs by a durable mask during dry
etch processing. More specifically, this invention
relates to the use of a highly durable etch mask to
protect desired regions of GaAs during the fabrication
of features such as waveguides, channels, facets,
mesas, and mirrors by dry etch processing in chlorine
10 containing ambients.

Background Art

Fabrication of opto-electronic devices in
GaAs requires the use of etch processes to produce
features for both microoptical and microelectronic
15 components. GaAs as used herein includes all
compounds, crystalline and polycrystalline, containing
gallium and arsenic with or without additional
elements. Examples of features which are transferred
into GaAs by etch processes include components such as
20 waveguides, channels, facets, mesas and mirrors. To
prepare components of this type having small, often
sub-micron size dimensions, etch processes which
provide a high degree of anisotropy are required. Wet
etch techniques are, in general, unsuitable. Etch
25 rates in wet processes are either isotropic or
dependent on crystallographic orientation. Dry etch
techniques, on the other hand, can avoid the dependence
in rate (under optimum conditions) to crystallographic
orientation and can provide the anisotropy required.

30 The majority of dry techniques used to
provide an anisotropic etch in GaAs are ion-based
processes which utilize chemistry to provide some form
of reactive assistance. The most common techniques of
this type include: Reactive Ion Etching (RIE),
35 Reactive Ion Beam Etching (RIBE), and Ion Beam Assisted
Etching (IBAE) [also known as Chemically-Assisted Ion

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Beam Etching (CAIBE)]. The chemistry utilized by these techniques for reactive assistance enhances etch rates, forms volatile etch products, and minimizes damage to the GaAs surface by energetic ions, neutrals, and/or radicals. In the dry etch processing of GaAs, ambients containing chlorine (atoms, molecules, neutrals, radicals) have been found most useful for providing suitable reactive assistance.

Some form of mask is required to protect desired regions of a substrate when using dry techniques to etch GaAs. Durable masks are often of particular value. The term "durable" as used herein defines the resistance of the mask to erosion during an etch process. An ideal mask is durable to the extent that it will not erode or change form during an etch process. For a mask to exhibit significant durability in the dry etch processing of GaAs, the ratio of the etch rate of the GaAs to that of the mask, i.e., the selectivity of the etch, must be high. Masks of materials of low durability are unsuitable for several reasons. First, the edge quality of etch features decreases as mask thickness increases. Second, mask features of dimensions smaller than the thickness of the mask are unstable and can break away or shift position during processing. Finally, mask erosion especially of edges can redeposit mask material into unwanted regions and degrade overall etch quality and uniformity.

In the dry etching of GaAs using chlorine for chemical assistance, few materials are known which when applied as thin layers (ca. 0.1 micron) offer the durability to survive an etch of from a few to many microns. Metals such as nickel (with titanium underlayer) and chromium and salts such as aluminum fluoride and strontium fluoride have been used with varying degrees of success. Wet techniques are

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frequently required to follow the dry etch processes to effect complete removal of these materials. What are needed are masks of high durability which are convenient to apply, pattern, and remove.

5 Disclosure of the Invention

An object of this invention is to provide a durable mask for the dry etch processing of GaAs in chlorine containing ambients. Quite unexpectedly, we have discovered that a mask formed of amorphous carbon
10 is highly resistant to attack by chlorine during dry etch processing in chlorine containing ambients. The amorphous carbon can be applied and removed by dry processing techniques.

A feature of this invention is that no wet
15 etch steps are required following lithographic patterning.

Another feature is that the amorphous carbon mask facilitates the fabrication of features such as waveguides, channels, facets, mesas, and mirrors by dry
20 etch processing in chlorine containing ambients.

Brief Description of the Drawings

Figures 1 (a-k) are sectional views illustrating various steps in the preferred method of forming and removing the highly durable mask resistant
25 to attack by chlorine.

Mode of Carrying Out the Invention

A mask of amorphous carbon can be prepared by a plasma-assisted CVD technique, i.e., by the decomposition of methane in an rf plasma. Other
30 hydrocarbons may be substituted for methane and a wide range of process conditions may be used. Alternative techniques for depositing this material may be used as well. These include variations of the plasma-assisted CVD technique, glow discharge (rf and dc) sputter
35 deposition, single and dual ion beam sputter deposition, reactive ion beam deposition, evaporation,

and ion plating. Depending on the deposition technique selected, the chemical and physical properties of the carbon mask may vary; the chemical properties from amorphous carbon (a-C) to hydrogenated amorphous carbon (a-C:H), the physical properties from diamond-like to graphite-like.

Amorphous carbons which serve as highly durable masks are determined easily through routine examination using, for example, scanning electron microscopy (SEM), and experiments such as those described in subsequent Examples I and II. To be considered highly durable, the ratio of the etch rate of GaAs to that of the mask should exceed 10:1 and more typically, 25-50:1. The ambient under which this durability is established may include chlorine containing gases or gas mixtures other than those given by the examples, e.g. BCl₃, BCl₃/Ar, etc., as long as the ambient is a suitable ambient for the dry etching of GaAs.

As shown in Figure 1a, a layer of amorphous carbon 12 is deposited on a substrate 10 of GaAs which may or may not contain an overcoating 11 of a material such as SiO₂, Si₃N₄ or amorphous silicon. An amorphous material is preferred for layer 11 to prevent transfer of grain boundary features. This overlayer, when present, provides an adhesion layer for the amorphous carbon and protects the surface of the GaAs from damage by ion bombardment during the deposition of the amorphous carbon. As shown in Figure 1b, a thin layer 13 of a material such as SiO₂, Si₃N₄, or amorphous silicon, is applied over the amorphous carbon layer. Again, an amorphous material is preferred for this layer to prevent transfer of grain boundary features. A layer 14 of a lithographically patternable material, e.g., photoresist or an electron beam sensitive resist, is then applied to the surface layer 13 as shown in

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Figure 1c. Layer 14 is then exposed and developed according to accepted practices to provide a pattern as shown in Figure 1d. Next a Reactive Ion Etch (RIE) step is used with an appropriate gas chemistry to selectively etch layer 13 through the exposed regions of the pattern in layer 14 as shown in Figure 1e. Alternative dry etch techniques may be substituted for any of the RIE steps specified in this description whenever convenient. If layer 13 is composed of SiO_2 or amorphous silicon a freon/oxygen mixture such as CF_4/O_2 (96 vol% CF_4 , 4 vol% O_2) can be used to selectively remove this layer by RIE. Another dry etch by RIE is then performed using oxygen (O_2) to selectively transfer the mask pattern in layer 13 through the amorphous carbon layer 12 to provide openings. This etch may also be used to remove the lithographically patternable layer 14 (Figure 1f). The sample is then dry etched by RIE using a suitable selective gas to remove the remaining layer 13 pattern (Figure 1g). By proper choice of materials this RIE step can also be useful for patterning layer 11 when present (Figure 1b). At this point, the processing of the amorphous carbon mask with patterned openings which expose the GaAs substrate is complete and the wafer is ready for dry etching in a chlorine containing ambient. Following transfer of the pattern into GaAs (Figure 1i), the amorphous carbon layer 12 is removed by RIE in O_2 (Figure 1j). Finally, the adhesion/protection layer 11 when present is removed by RIE under the same conditions used during pattern transfer (Figure 1k).

EXAMPLES

Two examples are presented here to illustrate the durability of amorphous carbon masks in the dry etch processing of GaAs in chlorine containing ambients using RIE and CAIBE. For convenience reasons only, the means used to pattern the amorphous carbon in these

examples deviate from the means described in the preferred embodiment. Also, the anisotropy of the etch was of no concern in the first example. RIE conditions are presented which do not optimize this aspect.

- 5 I. In the first example an amorphous carbon layer of thickness 1.38μ was deposited onto an evaporated SiO_2 adhesion/protection layer of thickness 0.34μ on a GaAs substrate. The amorphous carbon was deposited by the plasma-assisted CVD technique onto an
- 10 rf powered electrode at a self-bias voltage of 1600V and a methane pressure of 16 mTorr. Photoresist was applied and patterned for application of an Imidazole-based photoresist lift-off process. After preparation of the appropriate photoresist profile for the lift-off
- 15 process aluminum was deposited to a thickness of 0.1μ over the profile. The resultant lift-off pattern of aluminum on the amorphous carbon surface was then transferred through the amorphous carbon layer by RIE at 300 W in O_2 at a pressure of 40 mTorr. RIE was
- 20 again used to transfer the pattern through the SiO_2 adhesion/protection layer at 300 W in CF_4/O_2 (96 vol% CF_4 , 4 vol% O_2) at a pressure of 100 mTorr. A wet etchant was next used to strip the aluminum from the sample. Features to depths of 13.3μ were dry etched
- 25 through openings in the amorphous carbon and SiO_2 mask layers into the GaAs substrate by RIE at 330 W in Cl_2/Ar (25 vol% Cl_2 , 75 vol% Ar) at a pressure of 40 mTorr. Under these conditions the selectivity, i.e., the ratio of etch rate of the GaAs to that of the
- 30 amorphous carbon, was found to exceed 100/1 (109/1 actual). RIE at 300 W in O_2 at 40 mTorr was used to remove the remaining amorphous carbon and RIE at 300 W in CF_4/O_2 (96 vol% CF_4 , 4 vol% O_2) at 100 mTorr was used to remove the SiO_2 adhesion/protection layer.

II. In the second example an amorphous carbon layer of thickness 0.17μ was deposited onto an SiO_2 adhesion/protection layer of thickness 0.27μ on a GaAs substrate (actually a GaAs/AlGaAs double heterostructure containing substrate). The amorphous carbon was deposited by the plasma-assisted CVD technique, this time at a self-bias voltage of 400V and a methane pressure of 1 mTorr. The same procedure which was followed to pattern the amorphous carbon in the first example was followed to pattern the amorphous carbon in this example. Using the CAIBE technique, features to depths of 8.46μ were dry etched through openings in the amorphous carbon and SiO_2 mask layers into the GaAs/AlGaAs double heterostructure substrate using 500V Ar ions at a current density of 0.4 ma/cm^2 . The flow of Ar into the Kaufman type ion source was 3 sccm. Cl_2 gas was directed at near normal incidence to the sample during etching at a flow of 12.5 sccm. The chamber pressure during etching was maintained at $6 \times 10\text{E-}5$ Torr. The base pressure of the chamber before etching was allowed to reach $5 \times 10\text{E-}7$ Torr. The ratio of the etch rate of the GaAs (including the GaAs/AlGaAs double heterostructure) to that of the amorphous carbon in this example was greater than 90/1. The amorphous carbon and the SiO_2 adhesion/protection layers were both removed by RIE as described above for the first example. The anisotropy achieved during this etch was excellent.

Claims:

1. A method of forming a highly durable mask and using such mask in the dry etch processing of GaAs, comprising the steps of:

5 (a) providing a layer of amorphous carbon over the GaAs which is resistant to attack by chlorine on GaAs;

(b) forming openings in the amorphous carbon layer to expose the GaAs layer by dry processing to
10 form an amorphous mask;

(c) dry etch processing of the GaAs in chlorine containing ambients through the opening in the mask; and

(d) removing the mask.

15 2. The method of claim 1 wherein the amorphous carbon contains incorporated hydrogen.

3. The method of claim 1 wherein the mask is removed by dry processing.

4. A durable mask for use in the dry etch
20 processing of GaAs by chlorine containing ambients, comprising:

a layer of amorphous carbon over the GaAs resistant to attack by chlorine and having openings which expose the GaAs so that it can be etched by
25 chlorine.

FIG. 1a

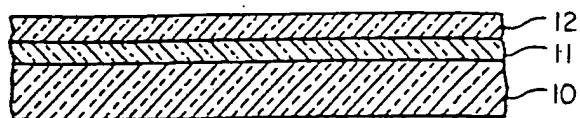


FIG. 1b

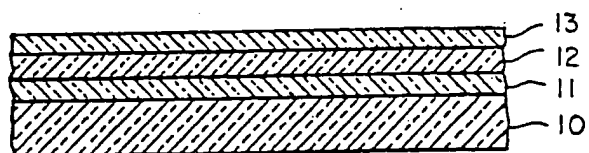


FIG. 1c

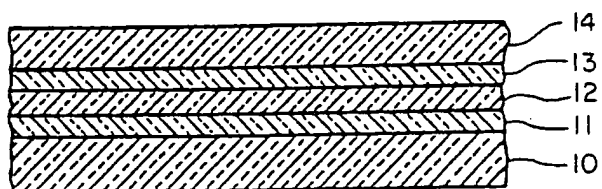


FIG. 1d

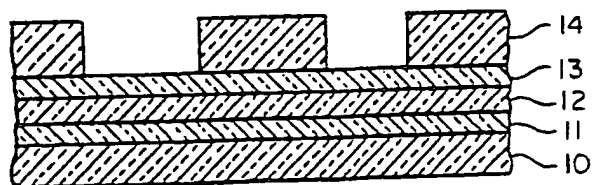


FIG. 1e

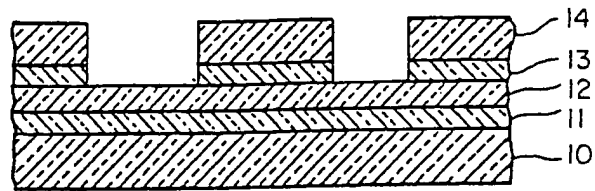


FIG. 1f

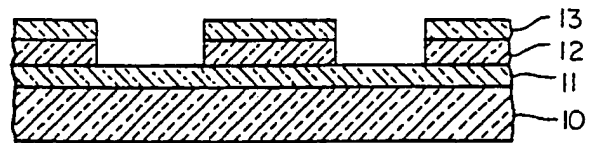


FIG. 1g

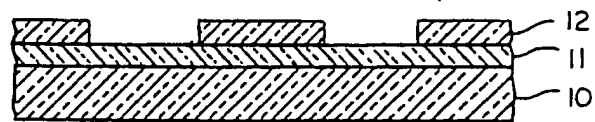


FIG. 1h

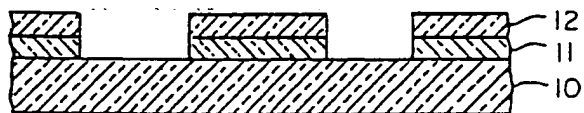


FIG. 1i

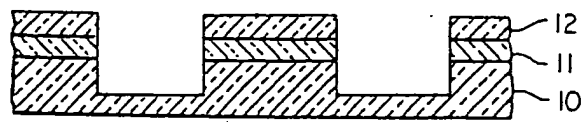


FIG. 1j

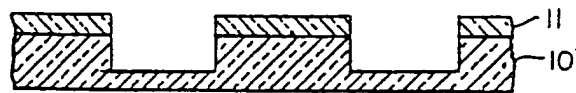


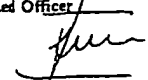
FIG. 1k



INTERNATIONAL SEARCH REPORT

PCT/US 92/02645

International Application No

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) ⁶		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl. 5 H01L21/308		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁷		
Classification System	Classification Symbols	
Int.Cl. 5	H01L	
Documentation Searched other than Minimum Documentation to the extent that such Documents are included in the Fields Searched ⁸		
III. DOCUMENTS CONSIDERED TO BE RELEVANT¹		
Category ⁹	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
Y	APPLIED PHYSICS LETTERS. vol. 48, no. 13, March 1986, NEW YORK US pages 835 - 837; KAKUCHI ET AL: 'Amorphous carbon films as resist masks with high reactive ion etching resistance for nanometer lithography' see the whole document ---	1,3,4
Y	EP,A,0 244 304 (VATUS ET AL) 4 November 1987 see abstract ---	1,3,4
A	PATENT ABSTRACTS OF JAPAN vol. 14, no. 455 (E-0985)28 September 1990 & JP,A,02 183 530 (MOTOKATA ET AL) 18 July 1990 see abstract --- -/-	1,3,4
<p>⁹ Special categories of cited documents : ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"&" document member of the same patent family</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report	
25 AUGUST 1992	2. 09. 92	
International Searching Authority	Signature of Authorized Officer	
EUROPEAN PATENT OFFICE	GORI 	

III: DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)		
Category *	Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to Claim No.
A	<p>NTIS TECH. NOTES. December 1986, SPRINGFIELD, VA US pages 1346 - 1347; ANONYMOUS: 'Masking technique for ion-beam sputter etching' see figure 1</p> <p>---</p>	1-4
A	<p>US,A,4 863 557 (KOKAKU ET AL) 5 September 1989 see example 2</p> <p>---</p>	1,3,4
A	<p>MATER. RES. SOC. 1991, SURFACE CHEMISTRY AND BEAM-SOLID INTERACTIONS SYMPOSIUM, BOSTON, MA, USA, 26-29 NOV. 1990 pages 129 - 133; HOWARD ET AL: 'Reactive ion etching of copper with SiCl₄ and CCl₂F₂' see page 130, paragraph 2 -paragraph 3</p> <p>---</p>	1,3,4
P,X	<p>US,A,5 022 959 (ITOH ET AL) 11 June 1991 see abstract</p> <p>---</p>	1-4

ANNEX TO THE INTERNATIONAL SEARCH REPORT
ON INTERNATIONAL PATENT APPLICATION NO.

US 9202645
SA 59658

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on
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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0244304	04-11-87	FR-A- 2598256 US-A- 4742026	06-11-87 03-05-88
US-A-4863557	05-09-89	JP-A- 63297435	05-12-88
US-A-5022959	11-06-91	JP-A- 2058221 US-A- 5102498	27-02-90 07-04-92

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